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Answer in Inorganic Chemistry Question for Joshua Q&A 147047

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Answer to Question #147295 in Inorganic Chemistry for shapopi phellep 2020-11-27T02:22:14-0500. Answers > Chemistry > Inorganic Chemistry. Question #147295. Given the following data: 2 H2 (g) + C (s) → CH4 (g) ΔG° = -51 kJ 2 H2 (g) + O2 (g) → 2H2O (l) ΔG° = -474 kJ ...

Answer in Inorganic Chemistry Question for shapopi phellep ...

Expert's answer. 2020-10-30T06:46:23-0400. The answer is c; ClO 2 (chlorine dioxide) Chlorine dioxide, ClO 2 reacts with hot water to produce chloric acid (HClO 3) and hydrogen chloride (HCl). 6ClO 2 (s) + 3H 2 O (l) 5HClO 3 (aq) + HCl (aq) The reaction takes place slowly and requires light energy to be initiated.

Answer in Inorganic Chemistry Question for Roshelle Q&A 141449

Answer to Question #147409 in Inorganic Chemistry for park sparrow 2020-11-27T10:21:05-0500. Answers > Chemistry > Inorganic Chemistry. Question #147409. how to use Molecular orbital theory to determine the hybridization of the metal centre in the complex. 1. [RhCl3(OH)3]3-2. [Cr(en)2(PMe3)2]

Answer in Inorganic Chemistry Question for park sparrow Q ...

Inorganic chemistry process. 1. Chemical conversion is dependent on reaction kinetics and in the case of reversible reaction on reaction equilibrium. Assess the statement by referring to ONE (1) inorganic process. 2. In the production of sulphuric acid, explain how each of the following conditions affecting the equilibrium and the rate of reaction.

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