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Atomic Structure Guided Notes Answers
Unit 3 Atomic Structure- Guided Notes
The Periodic Table ... - Add you answers from number 1 together - Round your answer to 2 decimal places • Practice: 1. A sample of Cesium is 75% ^{133}Cs , 20% ^{132}Cs , and 5% ^{134}Cs . What is the average atomic mass? 2. Carbon-12 is 98% abundant and Carbon-14 is 2% abundant. Name: Period: Unit 3 Atomic Structure- Guided Notes for

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Unit 1 - Atomic Structure 4.1 Defining the Atom I. Atomic Theory A. Modern Atomic Theory 1. All matter is made up of very tiny particles called atoms 2. Atoms of the same element are

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chemically alike 3. Individual atoms of an element may not all have the same mass. However, the atoms of an element

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The Periodic Table ... - Add you answers from number 1 together - Round your answer to 2 decimal places • Practice: 1. A sample of Cesium is 75% ^{133}Cs , 20% ^{132}Cs , and 5% ^{134}Cs . What is the average atomic mass? 2. Carbon-12 is 98% abundant and Carbon-14 is 2% abundant.

Name: Period: Unit 3 Atomic Structure- Guided Notes

Atom Notes Element- pure substance that cannot be separated into simpler substances by physical or chemical means. Represented by a chemical symbol Atom - Smallest particle into which an element can be divided and still be the same substance. Accepted scientific theory of atoms: 1. All

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substances are made of atoms. 2.

The Atom Notes

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ATOMIC STRUCTURE DIAGRAMS Step 1: Complete the squares for each element by adding the atomic number, name and atomic mass. - Write the atomic number at the top of the square. - Write the element's name under the symbol. - Write the atomic mass at the bottom of the square. Step 2: Determine the number of protons, neutrons, and electrons in

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Atomic Structure MODULE - 2 Notes
Atomic Structure and Chemical Bonding
chemistry has been defined as the study
of matter in terms of its structure,
composition and the properties. As you
are aware, matter is made up of atoms,
and therefore an understanding of the
structure of atom is very important. You
have studied in your earlier

ATOMIC STRUCTURE Notes

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Unit 3: Atoms & Elements - Chemistry by Mrs. Hinkson

atoms of different elements have different masses. (all. oxygen atoms have the same mass; and the mass of oxygen. is always different from every other element's mass) • Compounds contain atoms of more than one element. (ex: NaCl) • In a particular compound, atoms of different elements always. combine in the same way.

Chapter 4: Atomic Structure Notes Flashcards | Quizlet

Average atomic mass notes Homework:
1) Average atomic mass worksheet
Friday - 10/3: Activities: Today we took a pause from the math of average atomic masses and focused on the impacts of radioactive isotopes. An isotope is radioactive if it has an unstable nucleus. A stable nucleus will stay together.

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Unit 3 - Atomic Structure - Park Science

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ATOMIC STRUCTURE Notes The Atomic Structure Worksheet answers your questions by showing you how to structure your molecule, how to understand its properties and its structure.

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from 2.01 & 2.02 before attempting 2.02
assignment! 2.02 Guided Notes 2.03
Isotopes & Weighted Averages. 4.
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