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Oxygen gas . 1. The diagram in Model 1 illustrates a clipping of an aquatic plant in water. ... Provide specific evidence from Model 2 to support your answer. negatively - there are less bubbles with less light . Model 3 — Aquatic Plant Data ... variables that you think may change the outcome of the experiment, but since they are not being ...

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the fundamental unit of force $F = ma = (\text{kg})(\text{m@s}^{-2}) = \text{kg@m@s}^{-2} = \text{N}$ ii.

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For any sample of gas under ideal conditions, the relationship between the amount of gas in moles (n) and its temperature, pressure, and volume is given by the relationship. $PV = nRT$. in which R is the gas constant, with a value of $0.08206 \text{ L@atm/K@mol}$.

Chem 116 POGIL Worksheet - Week 1 Gas Laws - Part 1

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